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Key indicators

Single-crystal X-ray study
T = 293 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.046
wR factor = 0.146
Data-to-parameter ratio = 11.5

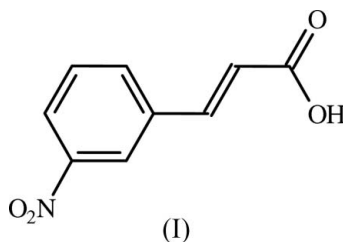
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

3-Nitrocinnamic acid

The title compound, $\text{C}_9\text{H}_7\text{NO}_4$, forms centrosymmetric dimers through intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds in the crystal structure. The nitro group deviates slightly from coplanarity with the benzene ring. The benzene ring and the carboxylic acid group are in an *E* configuration about the ethylenic double bond.

Comment

Various cinnamic acid derivatives form substrate intermediates with the enzyme papain (Huber, 1985). *m*-Nitrocinnamic acid crystallizes in two modifications and the unit-cell dimensions of these polymorphs have been reported previously (Schmidt, 1964). In this paper, we report the crystal structure of the β polymorph of *m*-nitrocinnamic acid, (I).



A perspective view of (I), with the atomic numbering scheme, is shown in Fig. 1. The bond lengths and angles agree well with literature values (Allen *et al.*, 1987). The $\text{C}1-\text{C}7-\text{C}8-\text{C}9$ torsion angle of $179.5(2)^\circ$ indicates that the benzene ring and the carboxylic acid group are in an *E* configuration about the $\text{C}7=\text{C}8$ bond and the propenoic acid moiety exists in an extended conformation. The alkenecarbonyl conforma-

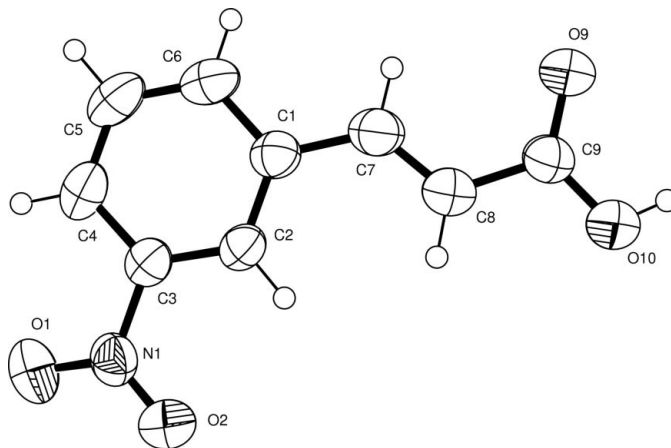


Figure 1

A view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small spheres of arbitrary radii.

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tion [C7—C8—C9—O9 = $-2.5(4)^\circ$] is synperiplanar, which is the most common conformation for *trans*-cinnamic acids (Leiserowitz, 1976).

The dihedral angle between the 3-nitro group and the benzene ring is $8.9(9)^\circ$. In a related structure, *viz.* *p*-nitrocinnamic acid (Kageyama *et al.*, 1993), the nitro group is coplanar (2.2°) with the benzene ring. With respect to the plane of the benzene ring, the 3-nitro group is oriented at an angle of 45.3° in 4-dimethylamino-3-nitrocinnamic acid (Huber, 1985), 3.6° in 3,5-dinitrocinnamic acid and 2.3° in the 3,5-dinitrocinnamic acid 2,5-dimethoxycinnamic acid complex (Desiraju & Sharma, 1991), 3.0° in the 3,5-dinitrocinnamic acid 4-(*N,N*-dimethylamino)benzoic acid complex and 6.1° in the 3,5-dinitrocinnamic acid 4-(*N,N*-dimethylamino)cinnamic acid complex (Sharma *et al.*, 1993).

The angle between the mean plane of the benzene ring and the mean plane of the propenoic acid moiety is $3.5(7)^\circ$ in (I) and 2.6° in 4-dimethylamino-3-nitrocinnamic acid (Huber, 1985). The corresponding angles in 4-chlorocinnamic acid (Glusker *et al.*, 1975), 4-iodocinnamic acid (Goud *et al.*, 1993), *p*-nitrocinnamic acid (Kageyama *et al.*, 1993), 3,5-dinitrocinnamic acid and the 3,5-dinitrocinnamic acid 2,5-dimethoxycinnamic acid complex (Desiraju & Sharma, 1991) are 14.1, 13.8, 4.7, 28.7 and 6.4° , respectively. In the 3,5-dinitrocinnamic acid 4-(*N,N*-dimethylamino)cinnamic acid complex, the propenoic acid group is twisted by 7.6° out of the mean plane of the benzene ring (Sharma *et al.*, 1993).

In the crystalline state, the molecules form O—H...O hydrogen-bonded dimers across an inversion centre (Table 1). These dimers are stacked along the shortest cell axis and lead to an $R_2^2(8)$ motif (Fig. 2) (Bernstein *et al.*, 1995).

Experimental

The title compound, (I), was prepared by dissolving *m*-nitrobenzaldehyde (6 g, 0.04 mol) and malonic acid (8.3 g, 0.08 mol) in a mixture of 5 ml of pyridine and 0.25 ml of piperidine. The solution was allowed to reflux for 1 h, with rapid evolution of CO₂. The resulting title compound was recrystallized from ethanol.

Crystal data

C ₉ H ₇ NO ₄	$D_x = 1.479 \text{ Mg m}^{-3}$
$M_r = 193.16$	Cu $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 25 reflections
$a = 3.7756(2) \text{ \AA}$	$\theta = 20\text{--}30^\circ$
$b = 9.4584(13) \text{ \AA}$	$\mu = 1.02 \text{ mm}^{-1}$
$c = 24.295(4) \text{ \AA}$	$T = 293(2) \text{ K}$
$\beta = 90.875(8)^\circ$	Block, colourless
$V = 867.52(18) \text{ \AA}^3$	$0.30 \times 0.20 \times 0.20 \text{ mm}$
$Z = 4$	

Data collection

Enraf–Nonius CAD-4 diffractometer	$R_{\text{int}} = 0.043$
ω – 2θ scans	$\theta_{\text{max}} = 67.9^\circ$
Absorption correction: ψ scan (North <i>et al.</i> , 1968)	$h = 0 \rightarrow 4$
$T_{\text{min}} = 0.751$, $T_{\text{max}} = 0.823$	$k = 0 \rightarrow 11$
1733 measured reflections	$l = -29 \rightarrow 29$
1478 independent reflections	3 standard reflections
1068 reflections with $I > 2\sigma(I)$	frequency: 120 min
	intensity decay: none

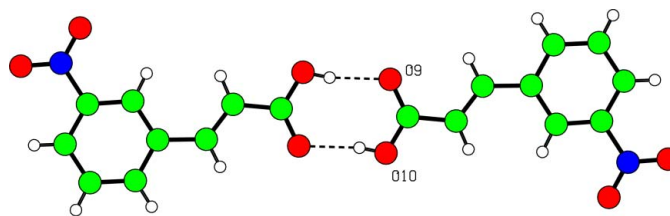


Figure 2

The O—H...O hydrogen-bonded (dashed lines) dimer in the structure of (I).

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0726P)^2 + 0.3649P]$
$R[F^2 > 2\sigma(F^2)] = 0.046$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.146$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.30 \text{ e \AA}^{-3}$
1478 reflections	$\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$
129 parameters	Extinction correction: <i>SHELXL97</i>
H-atom parameters constrained	Extinction coefficient: 0.0054 (11)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
O10—H10...O9 ⁱ	0.82	1.83	2.636 (3)	169

Symmetry code: (i) $-x + 1, -y - 1, -z$.

All the H atoms were placed in idealized positions (C—H = 0.93 \AA and O—H = 0.82 \AA) and constrained to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{parent atom})$.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *MolEN* (Fair, 1990); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

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